GUJCET - 3 (PCM) (ENGLISH MEDIUM)

DATE: 31.03.2023

Subject: Physics

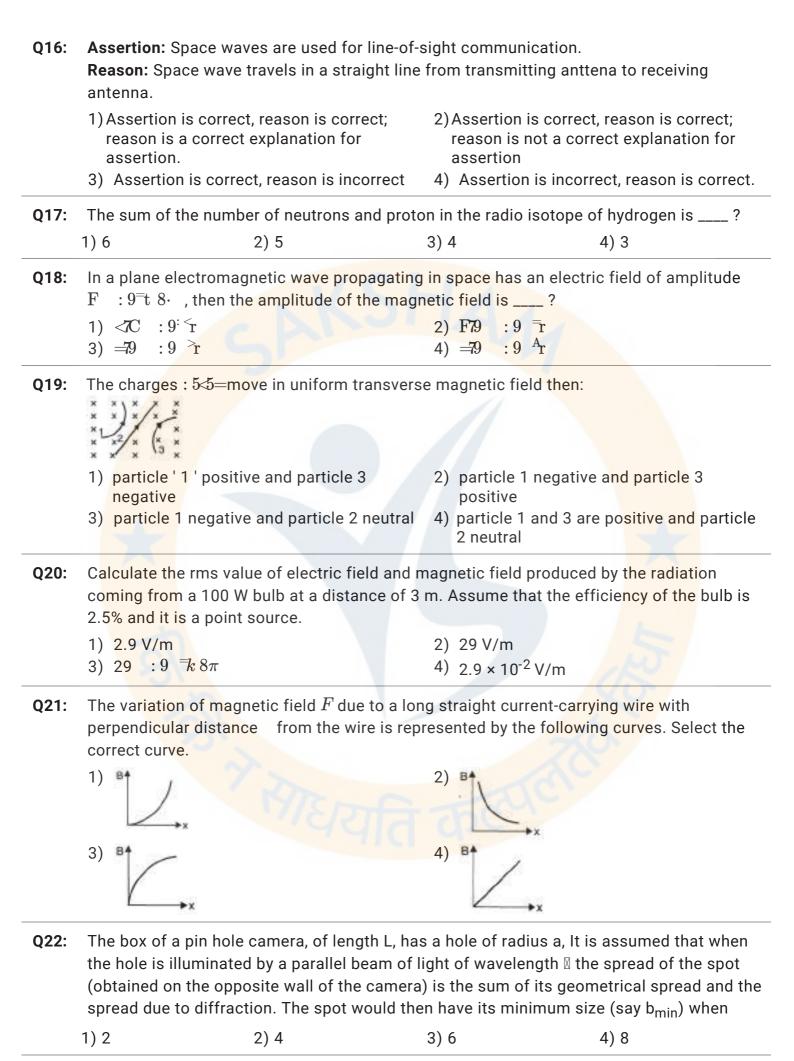
Physics

Q1:	In a photoelectric experiment the stopping potential for the incident light of wavelength >999 is 2 volt. If the wavelength be changed to $=999$, the stopping potential will be?			
	1) 2V	2) Zero		
	3) Less than 2 V	4) More than 2 V		
Q2:	A $rac{conductor}{has} : \!\!\! > \!\!\!\! > \!\!\! > : 9^{-:} \mathrm{FV}$ positive char	arge. The conductor has ?		
	1) 9 electrons in excess3) 9 electrons in short	2) 27 electrons in excess4) 27 electrons in short		
		· ·		
Q3:	Ratio of intensities of two waves are given waves is?	by4: 1. nen ratio of the amp <mark>litudes of t</mark> he two		
	1) 2: 1 2) 1 : 2	3) 4: 1 4.		
Q4:	are give <mark>n as</mark> :	et and the current in the circuit I at any instant which of the following holds good : ?		
	1) f M : V	2) f N : V		
	3) f L : V	4) none		
Q5:	A ray of light travelling inside a rectangular glass block of refractive index <is an="" angle="" at="" glass-air="" incidence="" incident="" of="" on="" surface="" the="">A . The refractive index of air is one. Under these conditions the ray will?</is>			
	 emerge into the air without any deviation 	2) be reflected back into the glass		
	3) be absorbed	4) emerge into the air with an angle of refraction equal to ${ m F9}$		
Q6:	Two parallel, long wires carry currents z ar	and $z_{<}$ with $z_{<}$ N $z_{<}$ 7When the current are in the		

same direction, the magnetic field at a point midway between the wire is $\mathrel{<\!\!\!\!< \!\!\!>} h$. If the

direction of z is reversed, the field becomes =9 h . The ratio z 8z is: ____ ?

	1) 4 3) 5	2) 3 4) :	
Q7:	A cell of internal resistance 3 ohm and emf 1 500 cm and resistance 3 ohm. The potential 1) $=9$ · t 8 $-$ 3) <9 · t 8 $-$	0 volt is connected to a uniform wire of length gradient in the wire is $ \begin{array}{c} \text{2)} : 9 \cdot \ t \ 8 \rightarrow \\ \text{4)} \ > \ t \ 8 \rightarrow \end{array} $	
Q8:	Assertion: Neutrons penetrate matter more r Reason: Neutrons are slightly more massive		
	 Assertion is correct, reason is correct; reason is a correct explanation for assertion. Assertion is correct, reason is incorrect 	 2) Assertion is correct, reason is correct; reason is not a correct explanation for assertion 4) Assertion is incorrect, reason is correct. 	
Q9:		ds any incident electromagnetic radiation but 2) >9g b 4) <9g b	
Q10:	wa <mark>velength o</mark> f 6000nm?	uctor with band gap of 2.8 eV. Can it detect a	
	1) yes 2) No	3) Both 4) None	
Q11:	The frequency of ac mains in India is? 1) $30 r8$ or R 3) $60 r8$ or R	2) 50 $r8$ or R 4) 120 $r8$ or R	
Q12:	The earth's magnetic field always has a verti 1) magnetic equator 3) geographic north pole	cal component except at t <mark>he ?</mark> 2) magnetic poles 4) latitude >A	
Q13:	Choose only false statement from the following	ing ?	
	1) In conductors the valence and	2) Substance with energy gap of the order	
	conduction band overlap 3) The resistivity of a semi conductor	of 10 eV are insulators 4) The conductivity of semiconductor	
	increase with increase in temperature	increase with increase in temperature	
Q14:	A positively charged particle moving due East directed vertically upwards. This particle will		
	1) move in a circular path with a decreased	2) move in a circular path with a uniform	
	speed 3) get deflected in vertically upward direction	speed4) move in circular path with an increased speed	
Q15:	ω equal resistors are first connected in series and then connected in parallel. What is the ratio of the maximum to the minimum resistance		
	1) ^ω	2) $\frac{:}{\omega^{<}}$	
	3) ω ^{<}	4) ÷ω	



Q23:	Find the angle of minimum deviation for an equilateral prism made of a material of refractive index 1.732. What is the angle of incidence for this deviation?				
	1) 60°, 30°	2) 30°, 60°	3) 90°, 60°	4) 60°, 60°	
Q24:		acitor has an electric fiel ite is : $ extit{H}5$ the force on ϵ			
Q25:				equilateral triangle of side p .	
		orce experienced by one		rest of the two is ?	
	1) $\frac{V}{p_{-}^{<}}$ 3) $\frac{V}{p_{-}^{<}}$		2) $\frac{\sqrt{V}}{p^{<}}$ 4) zero		
Q26:	In refraction, light w because in the seco	<mark>raves are</mark> bent on passin ond medium.	g from one <mark>mediu</mark> n	to second medium,	
	 Frequency is dif Coefficient of el 	ferent asticity is different	2) speed is diffe4) Amplitude is		
Q27:	fie <mark>ld at the ce</mark> ntre o	a steady current is bent f the loop is F . It is then e centre of this coil of ω	bent into a circula	o of on <mark>e turn. The</mark> magnetic r coil of ω turns. The	
Q28:	_	a coil with ${ m h}$ turns, all oncy in a magnetic field		and total $\frac{1}{1}$ resistance f , ralue of $\frac{1}{1}$ alue of $\frac{1}{1}$	
	1) N.A.B.R. 3) N.A.B.R.		2) N.A.B.4) N.A.B.		
Q29:		cident light falling on a perential		al plate is doubled, the	
	 double the earlie more than doub 		2) unchanged4) less than do	ubled	
Q30:	Magnetic lines of fo	rce due to a bar magnet	do not intersect be	ecause ?	
	field		repel each ot		
	3) the lines always force	diverge from a single	4) the lines need made to inter	d magnetic lenses to be est	
Q31:	Electrons in the ator	n are held to the nucleus	s by ?		
	 coulomb's force vander waal's fo 	rce	2) nuclear force4) gravitational		
Q32:	In a p-n junction did	de acting as a half-wave	e rectifier, which of	the following statements is	

- 1) The average output voltage over a cycle is non-zero
- 3) The depletion zone decreases in forward
- 4) The diffusion current increases due to

2) The drift current depends on biasing

Q33: In a discharge tube ionization of enclosed gas is produced due to collisions between -

- 1) negative electrons and neutral atoms/molecules.
- 2) photons and neutral atoms/molecules.
- 3) neutral gas atoms/molecules.
- 4) positive ions and neutral atoms/molecules.

Q34: A torque of : $9^{-A}h \cdot$ is required to hold a magnet at $F9^{-}$ with the horizontal component b of the earth's magnetic field. The torque to hold it at $=9^{-}$ will be ____ ?

1) A : $9^{-8}h$

biasing

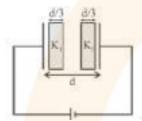
2) data is insufficient

forward biasing

3) $\stackrel{\cdot}{=}$: 9 $^{\text{A}}\text{h}$.

4) $A = :9^{B}h \cdot$

Q35:



Two dielectric slab of dielectric constant K_1 and K_2 and of same thickness is inserted in parallel plats capacitor and $K_1 = 2K_2$. Potential difference across slabs are V_1 and V_2 respectively then :-

1)
$$V_1 = V_2$$

2)
$$V_1 = 2V_2$$

3)
$$2V_1 = V_2$$

4)
$$4V_1 = V_2$$

Q36: The magnetism of magnet is due to?

- 1) The spin motion of electron
- 2) Earth
- 3) Pressure of big magnet inside the earth
- 4) cosmic rays

Q37: The de-broglie wavelength of an electron and the wavelength of a photon are the same. The ratio between the energy of the photon and the momentum of the electron is

1) h

2) c

3) $\frac{1}{u}$

4) $\frac{1}{r}$

Q38: For previous objective, which of the following graphs is correct

1)

2)

3)

4)

Q39: In Davison-Germer experiment, an electron beam is incident on a crystal. The reflected beam consists of?

- 1) photons
- 3) -rays

- 2) protons
- 4) electrons
- Q40: Consider an electron of mass · and charge e moving around a nucleus of charge < in circular orbit of radius . The initial frequency of light emitted by the electron is given as

- 2) $\int_{-\infty}^{\infty} \frac{1}{\sqrt{2\pi}} \frac{$

Subject: Chemistry

Chemistry

- Q41: How many atoms of calcium will be deposited from a solution of $V \rightarrow V \oplus$ by a current of 25 milliamperes flowing for 60 seconds
 - 1) >7BE : 9^{: E}
 - :9:<
 - 3) >7BE

- 2) >7BE : 9: A
- 4) $> BE : 9^F$
- The major product of the following reaction is Q42:



- 1) a hemiacetal
 - 2) an acetal
- 3) an ether

- 4) an ester
- Q43: Aryl halide is less reactive than alkyl halide towards nucleophilic substitution because ----?
 - 1) Less stable carbonium ion
 - 3) Inductive effect

- 2) Due to large $r ext{ } r\mu$ bond energy
- 4) Resonance stabilization and hybridisation of *H* attached to halide
- **Q44:** Which species is not a pseudohalide?
 - 1) CNO⁻
- 2) RCOO-
- 3) OCN $^{-}$
- 4) NNN⁻
- Q45: Which of the following type of forces are present in nylon-6, 6?
 - 1) Van der walls" forces of attraction
- 2) Hydrogen bonding
- 3) Three dimensional network of bonds 4) Metallic bonding
- Statement I: A mixture of phenol and benzoic acid can be separated by extracting its Q46: etheral solution with aq. b pRHc = solution. Statement II: Phenol is a weaker base than carbonic acid.
 - 1)Both Statement I and Statement II are correct and Statement II is the correct explanation of Statement I
 - 3) Statement I is correct but Statement II is incorrect
- 2)Both Statement I and Statement II are correct but Statement II is not the correct explanation of Statement I
- 4) Statement II is correct but Statement I is incorrect
- Aspirin is an acetylation product of $\{VTo \mid lgr: FFE\}$ Q47:
 - 1) o-hydroxybenzoic acid
 - 3) π -hydroxybenzoic acid

- 2) o-dihydroxybenzene
- 4) -dihydroxybenzene

Q48:	Sulphur molecule is converted into sulphur ion, when it			
	1) Gains two electrons	2) Loses two electrons		
	3) Gains two protons	4) Shares two electrons		
Q49:	The catalyst used in Rosenmund reaction is	3 ?		
	1) z 8b ∞	2) l o8T- o i >		
	3) Raney Ni	4) Na in Ethanol		
	•	,		
Q50:	Which one of the following compounds is n			
	1) $h \rightarrow V V \oplus$	$2) h \to V \circ V \oplus$		
	3) $V_{>}Nt 1Hb 2_{\rm B}$	4) $V = Nt \ 1Hb \ 2_B$		
Q51:	Which one is liquid at room temperature?			
	1) Vb ₌ V⊕	2) V _{<} b _A V⊕		
	3) Vb i	4) V _b AT		
Q52:	The random or zig-zag motion of the colloi	dal partiolog in the dispersion modium is		
QJZ.	referred to as?	ual particles in the dispersion medium is		
	1) Electro-osmosis	2) Electrophoresis		
	3) Brownian movement	4) Tyndall effect		
Q53:	Wh <mark>ich one of</mark> the following will undergo me	eta-sustitution on monochlor <mark>ination?</mark>		
	1) Ethoxybenzene	2) Chlorobenzene		
	3) Ethyl benzoate	4) Phenol		
Q54:	W <mark>hen a fres</mark> hly precipitated substance is c	onverted into a colloidal solut <mark>ion with th</mark> e help		
	of <mark>a third su</mark> bstance, the process i known a	as ?		
	1) Coagulation	2) Peptization		
	3) Electrodispersion	4) Dialysis		
Q55:	Which of the following intermediate specie	s is/are formed in the reaction of CH ₂ = CH -		
Ť	COOH (Acrylic acid) with HBr to give 3-bror			
	OH OH OH OH	e OH ON E		
	(I) CH2 = CH — (OH (II) CH2 = CH — OH (III) CH2 = CH —	OH CH CH CH C		
	1) I and II	2) III and IV		
	3) II, III and IV	4) I, II, III and IV		
Q56:	Phenyl isocyanides are prepared from which	ch of the following reactions {VTo[lgr:FFF}		
	Rosenmund's reaction	2) Carbylamine reaction		
	Reimer-Tiemann reaction	4) Wurtz reaction		
057		,		
Q57:	Which of the following silver compounds fi			
	1) $CxH\mu$ 3) CxS	2) $S \propto T$		
	3) <i>CxS</i>	4) Cxb c $=$		
Q58:	Green chemistry means such reactions whi	ch ?		

	1) produce colour during re	actions 2)	reduce the use and hazardous chemic	•	
	are related to the depletion layer	on of ozone 4)	study the reaction		
Q59:		part purple colour to	nottory is 2		
Q39.	The metallic oxide which imp		-		
	 Copper oxide Lead oxide 	•	Chromium oxide Manganese oxide		
	•	<u> </u>			
Q60:	Dimethyl glyoxime reagent is of ?	s used as co-ordinat	ng reagent in the q	uantitative estimation	
	1) Copper 2) Pal	ladium 3)	Silver	4) Nickel	
Q61:	In the body centred tetragon number is equal to?	al cystal system for	ned by square shee	ets, the co-ordination	
	1) 6	2)	8		
	3) 12	4)	none of t <mark>hese</mark>		
Q62:	D) CHCl ₂ / NaOH 2) HCHO / Conc NaOH 3) HQ*				
	Th	e correct product is .	2		
			206		
	1)	2)	CO_OH *HOOM		
	3) cm/cm	4)	OM COOM +CKJON		
Q63:	In the catalytic oxidation of of $bhi = .$ This oxide is			sed in the preparation	
	1) $b < c_{\rm A}$	2)	b < c >		
	3) b c <	4)	b < c > hi		
Q64:	Only stable organic fuctiona			?	
	1) <i>HHµ</i> <	2)	$HR_{<}$		
	3) HF <	4)	R-NC		
Q65:	Methylphenyl ether can be o	btained by reacting			
·	1) phenolate ions and meth	•	methoxide ions a	nd bromobenzene	
	3) methanol and phenol	•	bromo benzene a		
Q66:	The E.A.N. of iron in { ●1Vh	$\{2_{ m B}\}^{=}$ is ?			
	1) 32 2) 35	3)	38	4) 41	
Q67:	Which of the following statements is not applicable to chemisorption				

	1) It is slow	,	It is irreversible		
	3) It is highly specific	4)	It is independent of temperature		
Q68:	Chemical formula of rust is				
	1)	2)			
	3) $Nt < c = R < c$	4)	FeO. $R_{<}c$		
Q69:	Ethyl alcohol is industrially prepared from ethy	ylen	e by ?		
	Permanganate oxidation		Catalytic reduction		
	3) Absorbing in $R_{<\!\!\!/} gc$ a followed by hydrolysis	4)	Fermentation		
Q70:	In the electrolytic cell, flow of electrons is from	n			
	1) cathode to anode in solution	2)	cathode to anode through external supply		
	 cathode to anode through internal supply 	4)	anode to cathode through internal supply		
Q71:	Aceton <mark>e reacts with</mark> iodine $1 { m c} {\sim} 2$ to form iodofo	rm	in the presence <mark>of</mark>		
	1) S <mark>odium iodide</mark>		Sodium hypoiodide		
	3) <mark>lodine</mark>	4)	None of the above		
Q72:	n-Butyl alcohol on dehydration forms β -butyle because of the rearrangement ?	ene	as the chief product. This happens		
	$^{1)}$ CH3-CH2-CH2- $HR_{<}$ to CH3- HR_{-}	2)	CH3-CH2-CH2- $HR_{<}$ to (CH3)2CH-		
	$CH3-CH2-CH2-IIII_{<10}$ CH3- $IIII_{-}$	Í			
	3) OH2 HP OH2 OH2 to (OH2)2 H	4)	$HR_{<}$		
	$^{3)}$ CH3- HR -CH2-CH3 to (CH3)3 H	.,	$(CH_3)_3$ H_{to} CH_3 – CH_2 – CH_2 – $HR_<$		
Q73:	Mat <mark>ch the Colu</mark> mn I and Column II.		A LOT		
	Column I Column II				
	(A) Impurity defect (p) h → with anionic sites F-centres (B) Metal excess defect (q) • i with • 4				
	(C) Metal deficiency defect (r) h $\!$	ne ca	ationic sites vacant		
	1) (A)(r); (B) \rightarrow (p); (C) \rightarrow (q)		$(A) \rightarrow (p); (B) \rightarrow (q); (C) \rightarrow (r)$		
	3) (A) \rightarrow (r); (B) \rightarrow (q); (C) \rightarrow (p)	4)	$(A) \rightarrow (q); (B) \rightarrow (p); (C) \rightarrow (r)$		
Q74:	Which of the following has lowest boiling poin	t			
	1) -nitrophenol	2)	$\frac{\pi}{\text{-nitrophenol}}$		
	3) ϕ -nitrophenol	4)	phenol		
Q75:	Among the following substances the lowest v	•	•		
	1) Water	,	Mercury Postified spirit		
	3) Kerosene		Rectified spirit		
Q76:	When phenyl magnesium bromide reacts with				
	1) Benzene	•	Phenol		
	3) t-butyl benzene	4)	t-butyl phenyl ether		

	 A promoter An autocatalyst 	2) A positive catalyst4) None of these				
Q78:	<u> </u>					
Q70.	CII, - L-OCH, NaBI, -	product formed in the following chemical reaction is :				
	1) OH III OH CH, CH, OH	2) CH ₂ -CH ₃ -OH				
	3) CH, CH, - OII	4) OH CH, - C-CCH,				
Q79:	Sulpha drugs are used for					
	 Precipitating bacteria Decreasing the size of bacteria 	2) Removing bacteria4) Stopping the growth of bacteria				
Q80:	${ m Vi}$ bond length in ${ m J}$ •1 ${ m Vi}$ ${ m 2}_{ m A}$?	onoxide is : 7: $<$ E A , then what is the value of				
	1) : 7: A 3) : 7C<	2) : 7: <e 4) : 7: :</e 				
	Subject:	Maths				
	Math	ns				
Q81:	A signal which can be green or red with probability >8 Aand : 8 Arespectively, is received by station C and then trasmitted to station F . The probability of each station receiving the signal correctly is $=8$ >. If the signal received at station T is given, then the probability that the original signal is green, is?					
	1) $\frac{=}{A}$	2) ^B _C 4) ^F _{<9}				
	3) <= □	4) ½				
Q82:	it from the origin is the sphere	c). The locus of the foot of the perpendicular to				
	1) $\stackrel{<}{_{\sim}}4$ $\stackrel{\sim}{_{\sim}}4$ $\stackrel{<}{_{\sim}}4$ $\stackrel{\sim}{_{\sim}}4$ $\stackrel{\sim}{_{\sim}}4$ $\stackrel{\sim}{_{\sim}}4$ $\stackrel{\sim}{_{\sim}}4$ $\stackrel{\sim}{_{\sim}}4$ $\stackrel{\sim}{_{\sim}}4$ $\stackrel{\sim}{_{\sim}}4$ $\stackrel{\sim}{_{\sim}}4$ $\stackrel{\sim}{}4$ $\stackrel{\sim}{_{\sim}}4$ $\stackrel{\sim}{_{\sim}}4$ $\stackrel{\sim}{_{\sim}}4$ $\stackrel{\sim}{}4$ $\stackrel{\sim}{}4$ $\stackrel{\sim}{}4$ $\stackrel{\sim}{}4$ $\stackrel{\sim}{}4$ $\stackrel{\sim}{$	2) $\stackrel{<}{4}$ $\stackrel{<}{4}$ $\stackrel{<}{}$ $\stackrel{>}{}$ $\stackrel{>}{$				
Q83:	Solution set of $\rightarrow^{:}<4\rightarrow^{:}=M_{\overline{>}}$ is					
	1) 95:	2) $\frac{1}{8}5$:				
	3) $\frac{1}{B}$ 5 :	4) <u>:</u> B				
Q84:	8<					
	1)3) Does not exist	2) = 4) 0				
Q85:	The value of $-$:) $ \frac{A}{=}$ [$4 \neq :$) $ \frac{A}{=}$ [<u> </u>				

In the titration between oxalic acid and acidified potassium permanganate, the

manganous salt formed catalyses the reaction. The manganous salt is ____?

Q77:

11	
11	_
,	_

3) $\frac{:9}{:}$

$$2) \frac{A}{=}$$

4) 9

Q86: If $M \bullet A \rightarrow B 25$ then $\frac{s}{s}$ is ____?

1) $\overline{:4}$

None of these

In the function v1 $2\,\mathrm{M} \, {<\over<\,4} \, {
ightarrow} \, {>\over\cdot} \, {}^{\cdot} \, {}^{\cdot}$ $5\,1$ 92 is continuous at each point of its domain, then Q87: the value of v192 is ____ ?

3) <8=

2):8=

4) :8=

If x + 4 |y| = 6 y, then y as a function of x is _____? Q88:

1) continuous at x = 0

2) derivable at x = 0

3) $\frac{s}{s} = \frac{1}{s}$ for all x

4) none of these

A bag contains 4 white and 3 red balls. Two draws of one ball each are made without Q89: replacement. Then the probability that both the balls are red is ____?

3)

4) $\frac{>}{C}$

Which of the following statements is correct? Q90:

1) Every L.P.P. admits an optimal solution.

2) AL.P.P. admits a unique optimal solution.

3) If a L.P.P. admits two optimal solutions, it has an infinite number of optimal solutions.

4) The set of all feasible solutions of a L.P.P. is not a convex set.

 $\frac{t}{1:4} s$ M Q91:

1) $\frac{t}{:4}$ 4 r

2) $\frac{t}{:4} 4 r$ 4) $\frac{t}{:4} 4 r$

3) $\frac{t}{\cdot 4}$ 4 r

Integrating factor of equation $1 < 4 : 2\frac{s}{s} + 4 < M < : is ____?$ Q92:

2) < 4:

4) None of these

Q93:

lf

 $M\varepsilon$, then the value of k is ____?

4

1) 2

2) 4

3)6

4) 8

=5

 $_{9}^{8<} \frac{\overline{}}{\overline{} 4 \overline{}} s M$ Q94:

1) 0

2) =

3) =

4) None of these

Q95: Minimise Ho M < 4 = , subject to constraints < 4 >

9 and

9.

1)	9
	,	_

1

 $2\,M\!=\!5$ then the vector a makes with b

If three unit vectors a, b, c are such that Q96: and c respectively the angles

1)
$$>9^{\phi}5E9^{\phi}$$

2)
$$>A^{\phi}5$$

3)
$$=9^{\phi}5 B9^{\phi}$$

2)
$$>A^{\phi}5>A^{\phi}$$

4) $F9^{\phi}5 B9^{\phi}$

The area bounded in the first quadrant between the ellipse $\frac{1}{18} + \frac{1}{18} = 1$ and the line 3x+ 4y = Q97: 12, is ____?

1)
$$6(\pi - 1)$$

3)
$$3(\pi - 1)$$

A number is chosen from first 100 natural numbers. The probability that the number is Q98: even or divisible by 5, is ____?

1)
$$\frac{=}{>}$$

3)
$$\frac{>}{\Lambda}$$

4)
$$\frac{1}{\Delta}$$

Q99:

1) one point

2) two points

3) three points

4) infinite number of points

Q100: Which one of the following limit does not tends to unity?

1)
$$f \neq \neq g$$
 $f \neq g$ $f \neq g$

2)
$$f \neq \emptyset$$
 $8 < \frac{1 - 2}{2}$
4) $f \neq \emptyset$ $9 = \frac{1 + 2}{2}$

3)
$$f \neq \neq 9$$
 $\frac{14}{14}$

4)
$$f \neq \neq 9$$

Q101: A body moves according to the formula $M: 4 \le$, where v is the velocity at time t. The acceleration after 3 sec will be (v in cm/sec)

1)
$$\Leftrightarrow r\pi \ 8 \bullet -$$

2) :
$$< r\pi \ 8 \ \bullet -<$$

3)
$$Br\pi 8 \bullet -$$

Q102: The probability that a non leap year will have 52 Fridays is ____ ?

2) 2/7

3) 5/7

4) 6/7

Q103: If $v \ 1 \ 2 \ \mathrm{M} \ \neq 1 \oplus \infty \ 2 \ \mathrm{and} \ \mathrm{M} \ v) \stackrel{<\ 4\ =}{=} \left[$, then $\frac{s}{s} \ \mathrm{M} \ \mathrm{M} \ \mathrm{M} \ \mathrm{M} \ \mathrm{M} \ \mathrm{M} \right]$

1)
$$\frac{F - 1 \oplus \infty 2}{1 = < 2^{<}}$$

3)
$$\frac{F \neq 0}{1 + 1} \oplus \infty = 0$$

4) None of these

Q104: If a, b, c are mutually perpendicular vectors of equal magnitudes, then the angle between the vectors a and 4 4 is $_{---}$?

Q107:	Differential coefficie	nt of •— : is?			
	1) — :		2)	: <	
	3) = :		4) —	<u>:</u>	
Q108:	-	pen is selected from ea		nother box containing 3 white po then the probability that both th	
	1) : <		2) := 4) :\frac{1}{A}	911	
	3) :		4) $\frac{1}{A}$	Ā	
Q109:		erminant equation (in x)	p p	3.50	
	The roots of the dete	emmant equation (iii x)	π π q	$\pi \pi M9$	
	1) $Mp5q$			M p5 q	
	3) M $p5q$			Mp5 q	
Q110:	The principal value o	$f \neq \frac{1}{2} \left\{ \neq \right\} \leq \left[\int is \right]$?		
	1) \(\leq \)	() = [0	2) <	<u><</u>	
	3) =		_	None of these	
0111:	If a and b are mutual	lly perpendicular vector	s. then 1	1 4 2 ^{<} M	
	1) 4	, perperiarea.	2)		
	3) $p^{<}$ $q^{<}$		4) 1	2<	
Q112:	A coin is tossed and dice shows 6 is		bability	y that the coin sh <mark>ows the he</mark> ad a	and the
	1) <u>÷</u>		2) <u>:</u>	<u>:</u>	
	3) :		4) 1		
Q113:		: < A		-AC.	
	The rank of the matr		?		
	1) 1 if $p \mathrm{MB}$		2) 2	2 if p M :	
	3) 3 if p M $<$		4) 1	$if\ p\ M\!>$	
Q114:	Which of the following	ng functions has its inve	erse	_?	
	1) $f: R \to R$, $f(x) = a^{x}$	(2) f:	$:R \to R$, $f(x) = x + x - 1 $	
	3) $f:R_0 \to R^+, f(x) =$	x	4) f:	:[π , 2 π] \rightarrow [-1,1], $f(x) = \cos x$	
Q115:	If $v1$ 2 M $_{\overline{:4}}$ for	$f5$ then $v192\mathrm{M}$			
	1) 0	2) 1	3) 2	4) 3	

2) <u>:</u> Ā 4) - <

Q106: The value of - $1 \rightarrow$ $^{:}1 \rightarrow <\!\!22$ is ____ ?

1) <u>:</u> Ā

3) - <

Q116: $\frac{s}{s}$ { ightarrow 1- 24 - 12} M

1) 0

2) 1

3) - 1

4) - 2

Q117: $_{9}$ - s M

1)

3) 2

2) 0

4) 1

Q118: Let $v1\ 2M$ $\neq 5$ $\sqrt{}$ 9 and $x1\ 2Mt$. Then $1x\phi v2\ 192$ is ____?

1) 1

3) 0

2) -1

4) None of these

Q119: From which of the following the distance of the point $1:5 \le 5 = 2$ is :9

1) Origin

2) x-axis

3) y-axis

4) z-axis

Q120: Which one of the following is not bounded on the intervals as indicated

1) v $M < \frac{1}{2}$ on 195 2

5 2

3) $| 1 \ 2 \ M \bullet$ on 195 2

4) μ 1 2 M \rightarrow \rightarrow < on 1

5 2